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Holt Chemistry 14 Atoms and Moles Section: Electron Configuration In the space provided, write the letter of the term or phrase that best answers the question. ____ 1. Which of the following atomic models could be referred to as the 'solar-system' model of the atom? a. Thompson's plum-pudding model b. Rutherford model c. Bohr model d. quantum model ____ 2. According to the Bohr model ...

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There are over 100 different elements, which are made up of atoms. Elements can be divided into metals and non-metals. Chemical symbols and formulae are used to represent elements and compounds.

Atoms, elements and compounds test questions - KS3

Chemistry. States of matter. Add to My Bitesize Add to My Bitesize. Revise; Test. Solids, liquids and gases test questions. 1. In which two states are the particles randomly arranged? Liquid and ...

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Revise atomic structure and the arrangement of elements in the periodic table as part of National 5 Chemistry.

Atomic structure test questions - National 5 Chemistry

The elements can be placed in the periodic table. The position of an element provides information about its properties. Most elements are metals, with different properties to those of non-metals.

The periodic table test questions - KS3 Chemistry Revision

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Holt Modern Chemistry Chapter 3 Test Answers Holt McDougal Modern Chemistry 2 Chapter Test Chapter Test A, continued ____ 3. The periodic law states that a. no two electrons with the same spin can be found in the same place in an atom. b. the physical and chemical properties of the elements are functions of their atomic number. c.

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