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Nuclear Chemistry: Crash Course Chemistry #38 Chapter 21: Nuclear Chemistry (Chem in 15 minutes or less) CHM

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Radiation and Radioactive

Isotopes Chapter 21—

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9

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Chemistry: Part 2 of 9

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*Chemistry: Part 9 of 9*

Carbon 13 NMR Spectroscopy

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*Equations Song Nuclear Half*

*Life: Calculations Nuclear*

Chemistry Part 2 - Fusion

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Chemistry #39 Einstein's

Proof of  $E=mc^2$  How to

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Calculate the Mass Defect

and Binding Energy Chapter

14 – Chemical Kinetics: Part

1 of 17 ~~#0 value of a~~

~~Nuclear~~

~~Reaction, Unit 8, Atomic~~

~~Nucleus, Chp 4, Nuclear~~

~~Reactions, Class 12, Physics~~

Half Life Chemistry Problems

- Nuclear Radioactive Decay

Calculations Practice

Examples

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Nuclear Chemistry, 'Alpha

Decay' by Dr. Anju Agrawal,

M.Sc. Pt-III, Chemistry

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## **Radiation Exposure FSC Physics Part 2, Chapter 21, Nuclear Physics**

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Chemistry. Modern Chemistry  
175 Nuclearchemistry CHAPTER  
21 REVIEW Nuclear Chemistry  
SECTION 4 SHORT ANSWER

Answer the following  
questions in the space  
provided. 1. Match each of  
the following statements  
with the process(es) to  
which they apply, using one  
of the choices below: (1)

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fission only (3) both  
fission and fusion

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Chapter 21 Nuclear Chemistry  
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for Cr-53 is  $\frac{29}{24} = 1.21$ ; for Mn-51, it is  $\frac{26}{25} = 1.04$ ; for Fe-59, it  
is  $\frac{33}{26} = 1.27$ .

Positron decay occurs when  
the

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nucleons. protons and  
neutrons. nuclide. An atom  
identified by the number of  
protons



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Transformations • Rutherford  
in 1919 performed the first  
nuclear transformation. •

The transmutations are  
sometimes represented by  
listing in order, the target  
nucleus, the bombarding  
particle, the ejecting  
particle and the product  
nucleus. • The above

equation becomes:  ${}_{7}^{14}\text{N} + {}_{2}^{4}\text{He} \rightarrow {}_{8}^{17}\text{O} + {}_{1}^{1}\text{H}$

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Radioactivity; 21.2:  
Patterns of Nuclear  
Stability; 21.3: Nuclear

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A nuclear fuel. A  
fissionable isotope must be  
present in large enough  
quantities to sustain a  
controlled chain reaction.  
The radioactive isotope is  
contained in tubes called

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fuel rods. A moderator. A moderator slows neutrons produced by nuclear reactions so that they can be absorbed by the fuel and cause additional nuclear reactions. A coolant.

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