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(32) Identify and define the

three major types of

chemical bonding. In ionic

bonding, large numbers of

oppositely charged ions join

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~~Review~~ Because of mutual electrical attraction. In covalent bonding, atoms join by sharing electron pairs. In metallic bonding, atoms ...

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Section Review 1 1. What is the main distinction between ionic and covalent bonding?  
Answer (A): Ionic bonding involves the electrical attraction between large numbers of anions and cations. Covalent bonding

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rays. out of gamma,  
ultraviolet, and visible  
light, \_\_\_\_\_ radiation has  
the longest wavelength and  
\_\_\_\_\_ radiation has the  
greatest energy. visible,  
gamma . what is the correct  
formula ...

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What is the glue that hold  
atoms together? lower. When  
atoms bond, they will seek ?  
energy states. octet rule.  
What is the term for the  
tendency of atoms to share  
or transfer electrons to  
obtain a stable



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~~Review~~ configuration of 8 valence electrons? false . T/F The octet rule only applies to the transition metals. lose. Atoms with fewer than ...

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questions. Terms in this set

(21) What is a molecular  
orbital? A molecular orbital  
is the region of space  
around the nuclei of

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Covalently bonded atoms where shared electron pairs are likely to exist.

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