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Molecular Orbital Theory, Bonding
& Antibonding MO, Bond
Order, Homonuclear Diatomic~~

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~~Molecules Hybridization of Atomic Orbitals, Sigma and Pi Bonds, Sp Sp² Sp³, Organic Chemistry, Bonding Chapter 9 part 1~~
Intext:-Q.no-10.7 (Solution) |
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Chemistry 2e | OpenStax. 1. The cutting edge of a knife that has been sharpened has a smaller surface area than a dull knife. Since pressure is force per unit area, a sharp knife will exert a higher pressure with the same amount of force and cut through

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Answers more effectively. 3.

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~~Answers~~ the metals of group 1A, 2A, and 3A lose electrons, they form cations with positive charges equal to their group number.

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CHAPTER 9-Mastering Chemistry.

d. strong electrolyte. c. weak electrolyte. e. the amount of ion that carries 1 mol.... e. saturated. In water, a substance that ionizes completely in solution is c.... In water, a substance that partially ionizes in solution is ca.... An

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Answers is...

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9.1. Write the formulas for the
following coordination compounds:

(i) Tetraamminediaquacobalt (III)
chloride. (ii) Potassium

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~~Answers~~
tetracyanonickelate (II) (iii) Tris
(ethane-1,2-diamine) chromium
(III) chloride. (iv) Amminebromido
chloridonitrito-N- platinatc (II)
(v) Dichloridobis (ethane-1
,2-diamine) platinum (IV) nitrate.

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~~Answers~~ Tyndall effect, suspension, collide are important for MCQs, and short questions.

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Answer: (a) Primary valency is
ionisable. Question 6. Ammonia~~

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Answers
acts as a very good ligand but ammonium ion does not form complexes because. (a) NH_3 is a gas while NH_4^+ is in liquid form. (b) NH_3 undergoes sp^3 hybridisation while NH_4^+ undergoes sp^3 hybridisation. (c) NH_4^+ ion does not have any lone pair of

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Answers

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~~Answers~~ comprises the questions given in the exercises at the end of the chapter. The in-text questions are mainly about the IUPAC naming of the chemical compounds.

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Answers Here we have given Plus One Chemistry Chapter Wise Questions and Answers Chapter Chapter 9 Hydrogen. Kerala Plus One Chemistry Chapter Wise Questions and Answers Chapter 9 Hydrogen. Very Short Answer Type Questions (Score 1) Question 1.

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~~Answer~~ Producer gas is a mixture of

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As we noted in Chapter 9.2 the heat released by a reaction carried out at constant volume is identical to the change in internal energy (ΔE) rather than the enthalpy change (ΔH); ΔE is related to ΔH by an expression that depends on the change in the number of moles

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of gas during the reaction. The difference between the heat flow measured at constant volume and the enthalpy change is usually quite small, however (on the order of a few percent).

~~Chapter 9.6: Calorimetry~~

Page 41/45

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