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55 Termsmbizzario17. Chapter 1  
mastering chemistry. Matter. Property.  
Atoms. Molecule. Anything that has mass  
and takes up space, composed of particle.  
Any characteristic of matter thst allows us  
to recognize and d. Basic submicroscopic  
particles, building blocks of life.

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operator. You will see a multiplication dot  
( $\cdot$ ) appear in the answer box. Do not use  
the symbol x. For example, for the  
expression  $ma$ , typing  $m \cdot a$  would be  
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staff for the objective or, for noticeably  
less moolah, hire a assistance to complete  
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undergraduate level, 32 percent of the  
degrees awarded were in areas of strategic  
importance to the local economy,  
including degrees in health sciences and  
STEM. 2. Associate Professor Michael  
Lufaso has been teaching General .  
Chemistry I since he started at the school  
in 2006. He has taught General Chemistry  
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chemistry. Matter. Property. Atoms.  
Molecule. Anything that has mass and  
takes up space, composed of particle. Any  
characteristic of matter that allows us to  
recognize and define. Basic submicroscopic  
particles, building blocks of life. chemistry  
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prepare for examinations. Every chapter begins with a list of objectives, stating exactly the skills to develop in a particular unit. Each objective corresponds to a section in the textbook Living Chemistry. Three kinds of questions are provided for each objective to check the student's understanding, namely, short answer (Study Questions), multiple-choice, and fill-in. The answers for all questions are provided at the end of the chapter. The opening chapters cover the SI units, composition of matter, chemical bonding, compounds, chemical change, gases, respiration, and water. The subsequent chapters deal with solutions, acids, bases, salts, nuclear and organic chemistry, oxygen derivatives and hydrocarbons, polymers, and other organic derivatives. This textbook also explores the chemistry of carbohydrates, lipids, proteins, enzymes, and energy and carbohydrate



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metabolism. The remaining chapters discuss the chemistry of vitamins, hormones, body fluid, drugs, and poisons. Undergraduate chemistry students will find this book invaluable.

The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon, as it forms around particles of a metal catalyst. Carbon nanotubes are a recently discovered allotrope of carbon. Three other allotropes of carbon—buckyballs, graphite, and diamond—are illustrated at the left, as is the molecule methane,  $\text{CH}_4$ , from which nanotubes and buckyballs can be made. The element carbon forms an amazing number of compounds with structures that follow from simple methane, found in natural gas, to the

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complex macromolecules that serve as the basis of life on our planet. The study of chemistry also follows from the simple to the more complex, and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement.

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chemical concepts and problem solving into clear, manageable pieces to ensure students stay on track and motivated throughout their first, and often only, chemistry course. With the newly revised 6th Edition, best-selling author Karen Timberlake and new contributing author MaryKay Orgill connect chemistry to real-world and career applications. Their goal is to help students become critical thinkers by understanding scientific concepts that will form a basis for making important decisions about issues concerning health and the environment and their intended careers. The new edition introduces more problem-solving strategies, more problem-solving guides, new Analyze the Problem with Connect features, new Try It First and Engage features, conceptual and challenge problems, and new sets of combined problems--all to help students develop the problem-solving skills they'll

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