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Properties MIXED
REVIEW SHORT

ANSWER Answer the
following questions in
the space provided. 1.

Match the four
compounds on the
right to their
descriptions on the
left. b an ionic
compound that is
quite soluble in water

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- (a) HCl is an ionic compound that is not very soluble in water
- (b) NaNO_3

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review. absolute zero.

universal gas

constant. Boyle's law.

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Charles's law: a temperature at which a gas theoretically occupies zero volume. $0.08206 \text{ L atm/mol K}$. at constant temperature, the volume of a given amount of gas is.... at constant pressure, the volume of a given amount of gas is d....

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SECTION 3 SHORT

ANSWER Answer the

following questions in

the space provided. 1.

a The notation for

sodium chloride,

NaCl, stands for one

(a) formula unit. (c)

crystal. (b) molecule.

(d) atom. 2. d In a

crystal of an ionic

compound, each

cation is surrounded

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molecules. (c)
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dipoles. (b ...

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105 Chapter Test

Name Class Date

Chapter Test A,

continued Use this

figure to answer

questions 7 and 8.

_____ 7. A solution

containing 35 g of Li

2SO_4 dissolved in

100 g of water is

heated from 10°C to

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90°C. According to information in the figure, this temperature change would result in a. an additional 5 g of Li₂SO₄ in ...

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