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I. Chemistry 1. What is the difference
between a homogeneous mixture and a

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heterogeneous mixture? 2. List 2 examples of a homogeneous mixture and 2 examples of heterogenous mixtures. 3. List 2 examples of a pure substance and 2 examples of a mixture. 4.

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dsomerville. Combo with 9-1 Describing
and Measuring Motion and 27 others.
Terms in this set (264) motion. the state in
which one object's distance from another
is ...

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STUDY. PLAY. Kinetic theory. a theory of the behavior of gases that assumes heat is a process of energy transfer and the internal energy of a gas is the total energy

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of its particles. boiling point. the temperature at which a heated liquid turns into gas.

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STUDY. PLAY. There are many different
ways to classify matter, but one way is to

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use only two categories for ALL matter.

All matter can be classified as either:

Mixtures or substances. A substance made of two or more elements that cannot be separated by physical means is called a:

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STUDY. Flashcards. Learn. Write. Spell.
Test. PLAY. Match. Gravity. Created by.
dsomerville. Combo with 9-1 Describing
and Measuring Motion and 27 others.
Terms in this set (264) motion. the state in
which one object's distance from another
is changing. reference point.

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1. Metallic bonds create materials with properties that make them polar 2. The electrons at the outer energy level of the atom are called valence electrons 3. What

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are the associated charges with the
subatomic particles?

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8th Grade Physical Science Final EXAM

Review Packet 1. Match the following vocabulary words with the correct

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definition on the right _____ Physical
Science _____ Period (of Pendulum) ...
The study of energy and matter F. Testable
idea based on background knowledge G.
The application of scientific knowledge

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Review Packet~~

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Physical Science Fall Final Study Guide

Author: Karen A Compton Last modified

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Guide

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Physical Science – 1st Semester – Final
Exam Study Guide – (2015-2016) 1.

Physical Science Basics a. What is the
formula for calculating density? _____ i.

What is the density of a cube that is 15 g
and has a volume of 3 cm³? _____ ii.

Compare the density of solid H₂O to
liquid H₂O, and explain what observable

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effect this has.

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Exam Study Guide ...~~

Physical Science Final Study Guide:

Definition. All answers are True: Term.

Definition. Metals contain metallic bonds:

Term. Definition. Alloys are mixtures of

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different metals: Term. Definition. In a
chemical reaction the mass before = the
mass after: Term. Definition. $C + O_2 \rightarrow CO_2$
CO₂ is a chemical equation:

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Flashcards

Physical Science Final Study Guide 2nd

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Semester. STUDY. PLAY. Chemical Property. any characteristic of a substance that allows a substance to undergo a change that results in a new substance. To observe some chemical properties, the original substance being observed must _____ into one or more different substances.

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Semester Flashcards ...~~

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Guide: Nutrition and Physical Activity 1.
Obesity disparities 2. Difference between
overweight and obesity 3. What each food
group provides the body 4. Anorexia and

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bulimia 1. Obese disparities: higher risk for blacks, hispanics and low socioeconomic status-Prevalence of obese adults have nearly tripled in past 20 years-More than 30% of adults obese now ...

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-Health Science ...~~

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Exam Study Guide (2012-2013) Physical
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English; Published: December 10, 2015;
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