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 $360 = 0.54 \text{ rad}$ $\sin(30.5^\circ) = 0.5075$ $\sin(31.5^\circ) = 0.5225$ $\sin(31^\circ) = 0.51$ $\cos(30.5^\circ) = 0.86163$ $\cos(31.5^\circ) =$
 0.85264 $\cos(31^\circ) = 0.86$ 15. Some elementary chemistry textbooks give the value of R, the ideal gas
constant, as 0.0821 l atm K⁻¹ mol⁻¹. a.

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Solutions to exercises 1B.1(b) The mean speed is $8RT/v_{\text{mean}} M$ The mean translational kinetic energy is $m 3RT/2$ $3kT/2 M$ The ratios of species 1 to species 2 at the same temperature are $E_k/v_{\text{mean},1}$
 $v_{\text{mean},2}$ (i) $1/2 m v_1^2/M_1$ $v_{\text{mean},H}$ $1/2 m v_2^2$ and $2 v_{\text{mean},Hg}$ 200.6 4.003 1/2 2 $m v_{\text{rms}} E_k$ 1 E_k 2 7.079
(ii) The mean translation kinetic energy is independent of molecular mass and depends upon temperature alone!

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